

X-ray crystal structure of di- μ -carbonyldicarbonyl-bis(η^5 -pentamethylcyclopentadienyl)diruthenium

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An X-ray diffraction study of $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ru}(\text{CO})(\mu\text{-CO})]_2$ was performed, revealing its dimeric structure and centrosymmetric *trans* conformation. The Ru—Ru bond length (2.758(1) Å) and the separation between the Ru atom and the plane of the $\eta^5\text{-C}_5\text{Me}_5$ ligand (1.923 Å) are greater than those in the Cp analog and in the isostructural Cp and C_5Me_5 complexes of iron; this effect is due to the s inductive effect of the Me groups.

Key words: X-ray diffraction study; ruthenium(II) η -complexes; the metal—metal bond; the inductive effect.

It is known that the pentamethylcyclopentadienyl η^5 -ligand Cp* differs essentially from the nonsubstituted cyclopentadienyl ligand (Cp) in its chemical behavior; in particular, the complexes with Cp* are substantially more stable.¹ Of interest is the manifestation of this difference in the geometry of the complexes of analogous structures, which differ only by the ligands Cp* and Cp. An X-ray diffraction study of $[\text{CpFe}(\text{CO})(\mu\text{-CO})]_2$ (**1a**)² and $[\text{Cp}^*\text{Fe}(\text{CO})(\mu\text{-CO})]_2$ (**1b**)¹ with regard to neutron diffraction study of **1a**² demonstrated that these compounds have nearly the same geometry. In this work, the crystal structure of $[\text{Cp}^*\text{Ru}(\text{CO})(\mu\text{-CO})]_2$ (**2b**) was studied in order to compare it with the known structure of $[\text{CpRu}(\text{CO})(\mu\text{-CO})]_2$ (**2a**).³

(from 2.735(2) to 2.758(1) Å and from 1.910 to 1.923 Å, respectively), whereas the terminal Ru—CO bonds retain their lengths within experimental error (1.855(14) and 1.863(8) Å). Because the CH₃ groups in **1b** and **2b** induce no significant steric hindrance, the observed differences are completely attributable to the inductive σ effect of these groups, which is also confirmed by the fact that the geometric characteristics of **2b** and $[(\text{C}_5\text{Me}_4\text{Et})\text{Ru}(\text{CO})(\mu\text{-CO})]_2$ (**3**)⁴ are actually the same. Note, however, that the Ru— μ -CO bond lengths in **2a** and **2b** are significantly different (average values are 1.988(13) and 2.052(7) Å, respectively), though the poor accuracy of the structural study of **2a** by the photographic method casts some doubt on this difference.

Results and Discussion

The molecular structure of **2b** is shown in Fig. 1; its geometric parameters are given in Table 1. The molecule occupies a special position in the inversion center of the unit cell, and, therefore, the Cp* and the terminal CO ligands have *trans* orientation with respect to the Ru—Ru' bond; the molecule has approximate (non-crystallographic) symmetry C_{2h} . The Ru atom is at the distance of 1.923 Å from the plane of the cyclopentadienyl ring (Cp*); the C(Cp)—C(Me) bonds are bent out of the ring plane in the opposite direction with respect to the metal atom by 2.8–6.0°.

A comparison of structures **1a** and **1b** indicates that when Cp is replaced by Cp*, the Fe—Fe bond and the Fe—Cp* distance are slightly lengthened (by 0.021 and 0.013 Å, respectively), whereas the Fe—CO bonds are virtually unchanged. Similarly, as one goes from **2a** to **2b**, the Ru—Ru and Ru—Cp* distances are lengthened

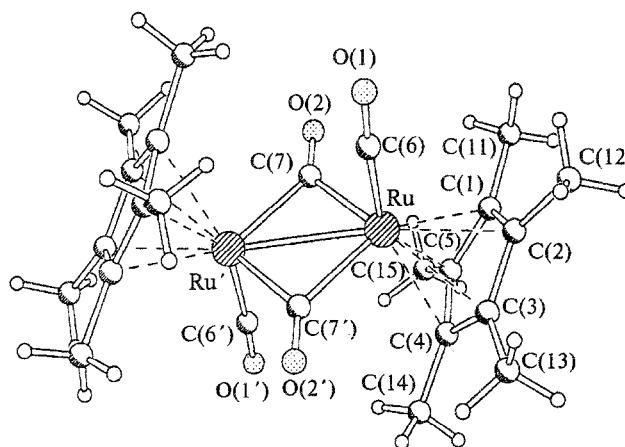


Fig. 1. The molecular structure of **2b**. Primed atoms are related to the basic atoms by an inversion center.

Table 1. Bond lengths and bond angles

| Bond | d/Å | Bond | d/Å |
|--------------------|----------|-----------------|----------|
| Ru—C(1) | 2.263(7) | O(2)—C(7) | 1.162(8) |
| Ru—C(2) | 2.240(7) | C(1)—C(2) | 1.44(1) |
| Ru—C(3) | 2.271(7) | C(1)—C(5) | 1.42(1) |
| Ru—C(4) | 2.303(7) | C(1)—C(11) | 1.52(1) |
| Ru—C(5) | 2.310(8) | C(2)—C(3) | 1.429(9) |
| Ru—C(6) | 1.863(8) | C(2)—C(12) | 1.50(1) |
| Ru—C(7) | 2.048(7) | C(3)—C(4) | 1.45(1) |
| Ru—Ru' | 2.758(1) | C(3)—C(13) | 1.48(1) |
| Ru—C(7') | 2.055(7) | C(4)—C(5) | 1.42(1) |
| Ru—Cp ^a | 1.924 | C(4)—C(14) | 1.49(1) |
| O(1)—C(6) | 1.14(1) | C(5)—C(15) | 1.49(1) |
| Angle | ω/deg | Angle | ω/deg |
| C(6)—Ru—C(7) | 91.6(3) | C(3)—C(2)—C(12) | 126.7(6) |
| C(6)—Ru—Ru' | 94.2(2) | C(2)—C(3)—C(4) | 107.5(6) |
| C(7)—Ru—Ru' | 47.8(2) | C(2)—C(3)—C(13) | 126.9(6) |
| C(6)—Ru—C(7') | 94.0(3) | C(4)—C(3)—C(13) | 125.3(6) |
| C(7)—Ru—C(7') | 95.5(2) | C(3)—C(4)—C(5) | 107.8(6) |
| Ru'—Ru—C(7') | 47.6(2) | C(3)—C(4)—C(14) | 125.5(7) |
| Cp—Ru—Ru' | 138.4(2) | C(5)—C(4)—C(14) | 126.2(7) |
| Cp—Ru—C(6) | 127.4(3) | C(1)—C(5)—C(4) | 108.7(6) |
| Cp—Ru—C(7) | 120.6(3) | C(1)—C(5)—C(15) | 124.5(7) |
| Cp—Ru—C(7') | 119.8(3) | C(4)—C(5)—C(15) | 126.8(7) |
| C(2)—C(1)—C(5) | 108.0(6) | Ru—C(6)—O(1) | 175.5(8) |
| C(2)—C(1)—C(11) | 125.4(7) | Ru—C(7)—O(2) | 137.9(6) |
| C(5)—C(1)—C(11) | 126.4(7) | Ru—C(7)—Ru' | 84.5(2) |
| C(1)—C(2)—C(3) | 108.1(6) | O(2)—C(7)—Ru' | 137.5(6) |
| C(1)—C(2)—C(12) | 124.9(6) | | |

^a Cp is the center of the cyclopentadienyl ring.

The coordination of the Cp* ligand in **2b** deviates significantly from symmetric η⁵ coordination, approaching pseudo-η³ coordination. The shortest distance between Ru and C(Cp*) (2.240(7) Å) is observed for the C(2) atom, which is in the pseudo-*trans* position with respect to the Ru—Ru' bond; the atoms C(1) and C(3), which are opposite Ru—CO (bridging) bonds, are far removed from the Ru atom (on the average, by 2.267 Å); whereas the average distance from Ru to C(4) and C(5), which are opposite the Ru—CO (terminal) bond, is 2.307 Å. Hence, the Ru—C(Cp*) distances increase in parallel with the π accepting ability of the pseudo-*trans* ligands.⁵ A similar effect is observed for complexes **1a**, **2a**, and **3**.

Experimental

The X-ray diffraction study was performed on an automated four-circle Siemens P3/PC diffractometer (graphite-monochromated Mo-Kα radiation) at room temperature. Red

Table 2. Fractional atomic coordinates (×10⁴ for O and C atoms and ×10⁵ for Ru atoms) and temperature factors (Å², ×10³ for O and C atoms and ×10⁴ for Ru atoms)

| Atom | x | y | z | U ^a |
|-------|-----------|-----------|---------|----------------|
| Ru | 570(5) | 10748(6) | 7302(3) | 225(2) |
| O(1) | 2950(6) | 2166(8) | 693(5) | 57(2) |
| O(2) | 995(6) | -2284(7) | 1231(4) | 44(2) |
| C(1) | -360(8) | 1175(9) | 2226(5) | 32(2) |
| C(2) | -207(7) | 2766(8) | 1889(5) | 28(2) |
| C(3) | -1309(7) | 3059(8) | 1124(5) | 28(2) |
| C(4) | -2141(7) | 1624(10) | 987(5) | 33(2) |
| C(5) | -1554(8) | 492(9) | 1674(5) | 33(2) |
| C(6) | 1862(8) | 1699(10) | 686(5) | 35(2) |
| C(7) | 576(7) | -1287(8) | 698(5) | 26(2) |
| C(11) | 537(10) | 419(13) | 3078(6) | 56(3) |
| C(12) | 838(10) | 3952(11) | 2344(6) | 48(3) |
| C(13) | -1645(9) | 4585(10) | 627(6) | 43(3) |
| C(14) | -3479(8) | 1465(12) | 338(7) | 51(3) |
| C(15) | -2107(12) | -1121(10) | 1836(8) | 59(4) |

^a 1/3 of the trace of the orthogonalized U_{ij} tensor.

crystals of **2b** (from THF) are monoclinic, *a* = 9.842(1), *b* = 8.384(1), *c* = 14.411(2) Å, β = 99.80(1)°, *V* = 1171.7(2) Å³, *Z* = 4 (C₂₄H₃₀O₄Ru₂), mol. weight 584.6, *d*_{calc} = 1.66 g cm⁻³, μ(Mo-Kα) = 13.2 cm⁻¹, F(000) = 588, space group *P*2₁/*n*. The intensities of 2411 independent reflections were measured using the θ/2θ scanning technique (2θ ≤ 54°), 2224 reflections with *I* > 2σ(*I*) were used in calculations. The structure was solved by the heavy-atom method and refined by the full-matrix least-squares method using the SHELXTL PLUS program package.⁶ Coordinates of all of the non-hydrogen atoms were refined anisotropically; all H atoms were refined using a riding model. The weighting scheme *w* = σ⁻²(*F*) was used. The final reliability factors are *R* = 0.045, *R*_w = 0.051, *S* = 3.27. Atomic coordinates are given in Table 2.

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References

1. R. G. Teller and J. M. Williams, *Inorg. Chem.*, 1980, **19**, 2770.
2. A. Mitschler, B. Rees, and M. S. Lehmann, *J. Am. Chem. Soc.*, 1978, **100**, 3390.
3. O. S. Mills and J. P. Nice, *J. Organomet. Chem.*, 1967, **9**, 339.
4. N. A. Baily, S. L. Radford, J. A. Sanderson, K. Tabatabaian, C. White, and J. M. Worthington, *J. Organomet. Chem.*, 1978, **154**, 343.
5. A. S. Batsanov and Yu. T. Struchkov, *Metalloorg. Khim.*, 1992, **5**, 53 [*Organomet. Chem. USSR*, 1992, **5**, 26 (Engl. Transl.)].
6. G. M. Scheldrick, *SHELXTL PLUS*, PC version, Siemens Analytical X-ray Instruments, Inc., Madison, Wisconsin, USA, 1989.

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